Carboxylic acids and Their derivatives

Self Evaluation Test -28

Which of the following is the weakest acid[CPMT 2001] 1.

(b) CH₃COOH СООН

(c) HCOOH



2. Pyruvic acid is obtained by [AFMC 1995]

- (a) Oxidation of acetaldehyde cyanohydrin
- (b) Oxidation of acetone cyanohydrin
- (c) Oxidation of formaldehyde cyanohydrin
- (d) None of these
- 3. The product obtained by dry distillation of calcium formate on reacting with ammonia yields
 - (a) Formamide
- (b) Acetamide
- (c) Acetaldehyde ammonia (d)

Urotropine

In the reaction 4.

 $C_8H_6O_4 \xrightarrow{\Delta} X \xrightarrow{NH_3}$

[Roorkee Qualifying 1998]

- The compound X is (a) Phthalic anhydride
- (b) Phthalic acid
- (c) o-xylene
- (d) Benzoic acid
- 5. Ethyl acetate reacts with CH_3MgBr to form

[MP PET 1999]

- (a) Secondary alcohol
- (b) Tertiary alcohol
- (c) Primary alcohol and acid (d)

Acid

- 6. In quick vinegar process of acetic acid, the temperature of mixture is [RPMT 2003]
 - (a) 300 K
- (b) 427 K
- (c) 500 K
- (d) 350 K
- Formic acid can reduce
- [CPMT 1987]
- (a) Tollen's reagent
- (b) Mercuric chloride
- (c) $KMnO_{4}$
- (d) All of these
- R 0.2 gm of fine animal charcoal is mixed with half litre of acetic acid (-SM) solution and shaken for 30 minutes

[BHU 1998]

- (a) The concentration of the solution decreases
- (b) Concentration increases

- (c) Concentration remains same
- (d) None of these
- Dimerisation in carboxylic acid is due to [KCET 2002] 9.
 - (a) Ionic bond
 - (b) Covalent bond
 - (c) Coordinate bond
 - (d) Intermolcular hydrogen bond
- colourless organic compound gives brisk effervescences with a mixture of sodium nitrite and dil. HCl. It could be

[CPMT 1978]

- (a) Glucose
- (b) Oxalic acid
- (c) Urea
- (d) Benzoic acid
- What is formed when benzovl chloride reacts with 11. aniline in presence of sodium hydroxide[BHU 1996]
 - (a) Acetanilide
- (b) Benzanilide
- (c) Benzoic acid
- (d) Azobenzene
- 12. Strong acid among the following is

[CBSE PMT 1992; AFMC 1998; BHU 2000]

- (a) CF_3COOH
- (b) CBr₃COOH
- (c) CH₃COOH
- (d) CCl₃COOH
- Aspirin is obtained by the reaction of salicylic 13. acid with

[AFMC 1998]

- (a) Acetone
- (b) Acetaldehyde
- (c) Acetyl chloride
- (d) Acetic anhydride

- Oxalic acid when reduced with zinc and H_2SO_4 gives

[Tamil Nadu CET 2001]

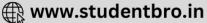
- (a) Glyoxallic acid
- (b) Glyoxal
- (c) Glycollic acid
- (d) Glycol
- A distinctive and characteristic functional group 15. in fats is

[NCERT 1981; MP PET 1995]

- (a) A ketonic group
- (b) An ester group
- (c) A peptide group
- (d) An alcoholic group







1360 Carboxylic acids and Their derivatives

(d) Nitrogen

Which substance will give amide when heated 16. with NH_3

17. Which acid has least pK_a value [CPMT 1982]

(a) $Cl_3C.COOH$ [CPMT 1997]

- (b) Cl₂CH.COOH
- (c) Cl.CH₂COOH (b) Hydrogen
- (d) CH₃COOH

(a) Potassium

(c) Ethane

Answers and Solutions

(SET -28)

- (a) Phenol is a weaker acid than carboxylic acids. 1.
- (a) $CH_3 C H + HCN \rightarrow CH_3 C H \xrightarrow{H_2O}$ 2. Pyruvic acid
- (d) $2(HCOO)_2Ca \xrightarrow{\text{Dry distillation}} 2HCHO + 2CaCO_3$ 3. $6HCHO + 4NH_3 \rightarrow (CH_2)_6N_4 + 6H_2O$
- 0 CONH₂
- O-MgBr(b) $CH_3 C - O - C_2H_5 + CH_3MgBr \rightarrow CH_3 - C - O - C_2H_5$

$$CH_{3} - C - CH_{3}$$

$$-Mg < OH \qquad CH_{3}$$

$$CH_{3} \qquad CH_{3}$$

$$CH_{3} \qquad CH_{3}$$

$$CH_{3} \qquad CH_{3}$$

$$CH_{3} - C - OH$$

- (a) $CH_3CH_2OH + O_2 \xrightarrow{Acetobacter} CH_3COOH + H_2O$
- (d) Tollen's 7. reagent $HCOOH + Ag_2O \rightarrow CO_2 + H_2O + 2Ag$ Fehling solution -

 $HCOOH + 2CuO \rightarrow CO_2 + H_2O + Cu_2O$

Mercury chloride –
$$2HCOOH + 2HgCl_2 \rightarrow 2CO_2 + 4HCl + 2Hg$$

- (a) Activated charcoal adsorbed the impurity of acetic acid by which the concentration of acetic acid solution decrease.
- (d) Intermolecular hydrogen bonding leads to 9. dimerisation of carboxylic acid in nonaqueous solvents.
- (c) $NaNO_2 + HCl \rightarrow HNO_2 + NaCl$ 10. $H_2NCONH_2 + HNO_2 \rightarrow CO_2 + NH_3 + H_2O + N_2$

CO₂ evolve with brisk effervescence.

(b) It is known as Schotten Baumann reaction. 11.

$$C_6H_5NH_2 + ClCOC_6H_5 \xrightarrow{NaOH} C_6H_5NHCOC_6H_5 + HCl \\ \text{Antiline} \qquad \text{Benzoyl chloride}$$

(a) Due to -I effect of three F atom CF_3COOH is 12. a strong acid.

OH
$$COOH$$
 $COOH$ $COOH$

14. (c)
$$COOH \atop COOH \atop COOH \atop Oxalic acid} + 4[H] \xrightarrow{CH} CH_2OH \atop H_2SO_4} |COOH \atop COOH \atop Clycolicacid} + H_2O$$

- (b) Fat is the ester of higher acids & glycerol. 15.
- (a) $2K + 2NH_3 \rightarrow 2KNH_2 + H_2$
- (a) $(Cl_3C-COOH)$ Trichloroacetic acid has least 17. pka value and is most acidic.





